CYCLOADDITIONS. XXV. CYCLOADDUCTS FROM DIPHENYL SULFIDE AND ALLYLIC CHLORIDES¹ John E. Baldwih² and Richard E. Peavy³ Department of Chemistry and Chemical Engineering University of Illinois, Urbana, Illinois 61801 (Received in USA 17 June 1968; received in UK for publication 2 September 1968) The allyldimethylsulfonium salts <u>1</u> react with alkyllithium reagents to give allylcarbinyl methyl sulfides <u>4</u>, ⁴⁻⁸ presumably through a Claisen-like rearrangement of the anion <u>3</u> involving 6 electrons disposed over 5 atomic centers.^{7,9}



This reaction hampers chemical investigations of alkylidenesulfuranes 2; the equilibrium between the sulfuranes 2 and 3 is displaced irreversibly through the rearrangement $3 \rightarrow \frac{1}{2}$ before intermolecular reactions of 2 can be effected.

To obviate such rearrangements, allyldiphenylsulfonium tetrafluoroborate (5) was prepared and found to behave as a normal vinyl ylide, exhibiting behavior best interpreted on the basis of tetravalent sulfur intermediates.⁶

Our experience with attempts to prepare allyldiphenylsulfonium tetrafluoroborate has been quite different. Diphenyl sulfide and allyl chloride in 1,2-dichloroethane gave on treatment with silver tetrafluoroborate a 25% yield of crude salt, mp 135-145°. Recrystallization from 3-pentanone gave an analytically pure sample of $C_{15}H_{15}BF_4S$, mp 149.5-150°; nmr (CDCl₃) τ 2.50

5029

(m, 9), 6.55 (m, 1), 8.23 (d, $\underline{J} = 7$ Hz, 2) and 8.65 (d, $\underline{J} = 7$ Hz, 3). No bands attributable to a C-C double bond were observed in the infrared spectrum. (By contrast, allyldimethylsulfonium bromide has infrared absorptions at 1640 and 955 cm⁻¹ in CHCl₃.) This salt was formulated as the cycloadduct <u>6</u>. Upon alkaline hydrolysis¹⁰ it gave an oil identified as the aromatic sulfide <u>7</u>; nmr τ 2.81 (m, 9), 3.11 (m, 1), 3.83 (m,1), and 8.16 (doublet of doublets, <u>J</u> = 1.5 and 6.5 Hz, 3); infrared band at 975 cm⁻² in CCl₄.



Methallyl chloride and diphenyl sulfide gave under similar conditions a sulfonium salt having nmr absorptions at τ 2.60 (m, 9) and three singlets at 6.55, 8.17, and 8.58 (2:3:3). Formula $\underline{\delta}$ is the tentative structural assignment.



The reactions leading to salts $\underline{6}$ and $\underline{8}$ may be postulated as proceeding from the allyldiphenylsulfonium salts through the Claisen rearrangement¹¹⁻¹³ product 2 to <u>6</u> and the intramolecular 1,3-cycloadduct <u>10</u> to <u>8</u>. Other mechanistic possibilities, of course, obtain.

Confirmation of structural assignments for $\underline{\delta}$ and $\underline{8}$ is now being sought through synthesis of anthentic samples.

References

- 1. Partial support of this work by the National Institutes of Health (GM 14381) is gratefully acknowledged.
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